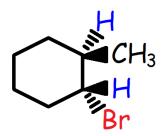
Organic Chemistry Exam 2 Worksheet

Organic Chemistry Tutor

1. Which of the following represents the name of the compound shown below?

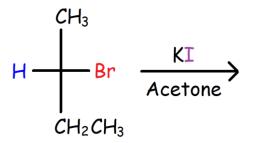


- A. (1R, 2R)-1-bromo-2-methylcyclohexane
- B. (1R, 2S) -1-bromo-2-methylcyclohexane
- C. (1S, 2R) -1-bromo-2-methylcyclohexane
- D. (1S, 2S) -1-bromo-2-methylcyclohexane

- 2. Which of the following alkyl halides would you expect to undergo a SN2 reaction most rapidly?
- A. 1-Bromobutane
- B. 2-Bromobutane
- C. 2-Bromo-2-methylbutane
- D. 2-Bromo-3-methylbutane

- 3. Which of the following solvents will work best in a SN1 reaction?
- A. Hexane
- B. Ethanol
- C. Acetone
- D. DMF
- E. H₂O

4. What is the major product of the following reaction?

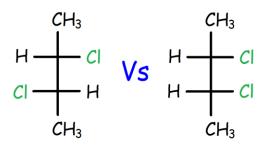


- A. (R)-2-Iodobutane
- B. (S)-2-lodobutane
- C. A racemic mixture of A and B
- D. None of the above

- 5. What is the major product that will be formed when 1-Butene reacts with HBr?
- A. 1-Bromobutane
- B. (R)-2-bromobutane
- C. (S)-2-bromobutane
- D. A racemic mixture of B and C

- 6. Which of the following reagents will convert 3-Methyl-1-butene to 3-Methyl-2-butanol?
- (A) 1. BH₃/THF 2. H₂O₂, OH⁻, H₂O
- (B) 1. Hg(OAc)₂, H₂O 2. NaBH₄
- (C) H_2O/H_2SO_4
- (D) 1. MCPBA 2. H₃O⁺
- (E) 1. OsO₄ 2. NaHSO₃, H₂O

7. What is the relationship between the two compounds shown below?



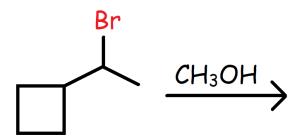
10. 3g of a chiral stereoisomer was dissolved in 20 mL of solution. The observed rotation measured by a polarimeter was found to be -9° at 25° C using a wavelength of 589 nm. The sample tube of the polarimeter is 20 cm long. What is the specific rotation of the stereoisomer?

- A. Enantiomers
- B. Diastereomers
- C. Constitutional Isomers
- D. Meso Compounds
- 8. Which of the following reagents will produce a meso compound after reacting with Cyclohexene?
- 11. How many possible stereoisomers exist for 3-Ethoxy-4-methylcyclohexanol?

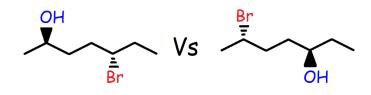
- A. Br₂ / CCl₄
- B. 1. MCPBA 2. H₃O⁺
- C. H₂O / H₂SO₄
- D. CH_3OH / H_2SO_4
- E. 1. OsO₄ 2. NaHSO₃, H₂O
- 9. Which of the following is the major product that is formed in the reaction of Propyne with Sodium Amide followed by Ethyl Bromide?
- A. 1-Pentyne
- B. 2-Pentyne
- C. 1-Pentene
- D. 2-Pentene

12. What is the major product that will be formed when 3,3-Dimethylcyclohexane reacts with Hydrobromic Acid?

13. What is the major substitution product of the reaction shown below?



16. What is the relationship between the two compounds shown below?



- A. Diastereomers
- B. Enantiomers
- C. Constitutional Isomers
- D. Meso Compounds

- 14. Which of the following reagents will convert 2-Butyne into Cis-2-butene?
- 17. Calculate the % enantiomeric excess for a solution that contains 16g of (R)-2-bromobutane and 4g of (S)-2-bromobutane.

- A. H_2/Pt
- B. H₂ with Pd/C
- C. $Na(s) / NH_3(I)$
- D. H₂ / Lindlar's catalyst

- 15. Which of the following compounds will be produced when 2,3-Dimethyl-2-butene reacts with Ozone followed by Dimethyl Sulfide?
- A. Ethanal
- B. Propanal
- C. Acetone
- D. 2-Butanone

- 18. Which of the following alkenes is most stable?
- A. 1-Hexene
- B. Trans-2-hexene
- C. Cis-2-hexene
- D. 2-Methyl-2-pentene
- E. 2,3-Dimethyl-2-butene

- 19. Which of the following alkyl halides will react most readily in a SN2 reaction?
- 22. Which of the following reagents can be used to convert 1-Methylcyclohexene to 1-Bromo-2-methylcyclohexane.

- A. CH₃CH₂-F
- B. CH₃CH₂-Cl
- C. CH₃CH₂-Br
- D. CH₃CH₂-I

- A. Br₂ / CCl₄
- B. Br_2/H_2O
- C. Br₂ / CH₃OH
- D. HBr
- E. HBr / ROOR

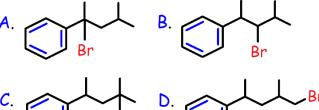
- 20. Which of the following represents the correct rate law expression for a SN1 reaction?
- 23. Which of the following reagents can be used to accomplish the transformation shown below?

- A. Rate = k[Substrate][Nucleophile]
- B. Rate = k[Substrate]
- C. Rate = k[Substrate][Base]
- D. Rate = k[Base]
- E. Rate = k[Nucleophile]

- $\bigcirc \longrightarrow \bigcirc \times_{c}^{c}$
- A. CH₂N₂ / heat
- B. $Zn(Cu) / CH_2I_2$
- C. CHBr₃ / KOH
- D. CHCl₃ / KOH
- E. Cl₂ / CCl₄
- 21. Which of the following statements is not true?
- A. The rate of a SN2 reaction will double if the concentration of the nucleophile doubles.
- B. The rate of an E1 reaction will triple if the concentration of the alkyl halide triples.
- C. The rate of an E2 reaction will increase by a factor of 6 if the concentration of the alkyl halide doubles and the concentration of the base triples.
- D. The rate of a SN1 reaction will quadruple if the concentration of the nucleophile quadruples.

- 24. Which of the following nucleophiles will yield the fastest SN2 reaction in a polar aprotic solvent such as DMSO?
- A. F
- B. Cl-
- C. Br⁻
- D. I

25. Which of the following alkyl halides will have the greatest reactivity in a SN1 reaction?



26. Which of the following molecules is optically inactive?

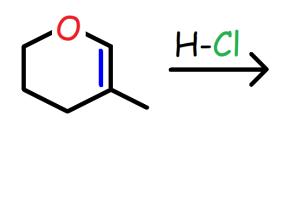
27. Which of the following solvent mixtures will yield the fastest reaction between Tert-butyl Bromide and Potassium Iodide?

- A. 25% H₂O / 75% CH₃OH
- B. 50% H₂O / 50% EtOH
- C. 50% H₂O / 50% MeOH
- D. 100% CH₃CH₂OH

28. Which of the following nucleophiles will work best in a SN2 reaction with a polar protic solvent such as EtOH?

- A. OH-
- B. CH₃S⁻
- C. CH₃SH
- D. SCOCH₃
- E. C₆H₅O⁻

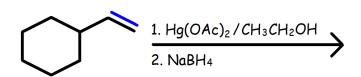
29. What is the major product of the reaction shown below?



30. Which of the following reagents will proceed with anti-addition upon reacting with

- 1-Methylcyclohexene?
- A. H_2/Pt
- B. KMnO₄, Cold / OH⁻, Dilute
- C. 1. BH₃ / THF 2. H₂O₂, OH⁻
- D. Br₂ / CH₃OH

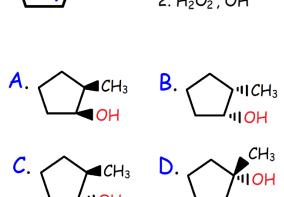
- 31. What is the major product of the reaction shown below?
- 34. Which of the following reagents will convert 1-Butyne to 2-Butanone?



- (A) H_2 / Pd
- (B) Li(s) / $CH_3NH_2(I)$ @ -50° C
- (C) 1. R₂BH 2. H₂O₂, NaOH
- (D) HgSO₄, H₂SO₄, and H₂O

32. Which of the following compounds could be a major product of the reaction shown below?

35. What is the major product that will be produced when 1 mole of Br₂ reacts with 1 mole of 2-Methylpropene and 2 moles of NaCl?



- 33. How many possible products can be produced from the reaction of 2-Pentene with HBr?
- 36. What is the major product that will be produced when 3-Methylcyclohexene reacts with BrCl?

- 37. Which of the following statements is not true?
- A. The SN2 mechanism is a concerted reaction where all bond breaking and bond formation processes occur simultaneously.
- B. The SN2 reaction proceeds with inversion of configuration.
- C. The SN1 reaction produces an unequal racemic mixture of products due to the presence of an intimate ion pair.
- D. The SN2 reaction is enhanced by a polar aprotic solvent because it solvates the cation but not the nucleophilic anion.
- E. The SN2 reaction generates carbocation intermediates and is subject to carbocation rearrangements.

- 39. What is the major product produced from the reaction of 1-Bromopentane with Methanol?
- A. 1-Methoxypentane
- B. 1-Pentene
- C. 2-Pentene
- D. (R)-2-Methoxypentane + (S)-2-Methoxypentane

- 38. Which of the following compounds will undergo solvolysis in aqueous Methanol most rapidly?
 - A. CIB. CIC. CID. CI
- 40. How many intermediates are present in the potential energy diagram for the reaction shown below?

$$\xrightarrow{\text{Br}} \xrightarrow{\text{H}_2\text{O}} \xrightarrow{\text{OH}}$$

- A. 2
- B. 3
- C. 4
- D. 5

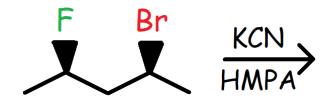
- 41. What is the major product of the reaction between 5-Bromo-2-methyl-1-pentanol with Sodium Hydride?
- 44. Which of the following is the major product of the reaction between 2-Bromopentane and Sodium Ethoxide in Ethanol?
- A. 1-Pentene
- B. Cis-2-pentene
- C. Trans-2-pentene
- D. 3-Pentene

- 42. What is the major product of the reaction between Cyclohexene with Br_2 in H_2O followed by NaH?
- 45. How many possible elimination products can be produced in the reaction of 2,2-Dimethylcyclohexanol with concentrated Sulfuric Acid and heat?

- 43. What is the major product of the reaction shown below?
- HO CH_2Cl_2
- 46. What is the major product of the reaction when 2-Bromo-2-methylpentane reacts with Sodium Methoxide in Methanol?

- 47. What is the major product of the reaction when 2-Bromo-2-methylpentane reacts with Potassium Tert-butoxide in Tert-butanol?
- 50. Which of the following bases will convert 1-Bromobutane to 1-Butene with the greatest yield?
- A. KOH
- B. NaOCH₃
- C. NaOCH₂CH₃
- D. KOC(CH₃)₃
- E. KOC(CH₂CH₃)₃

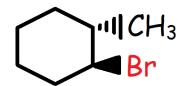
- 48. What is the major product of the reaction when 2-Fluoropentane reacts with Sodium Methoxide in Methanol?
- 51. What is the major product of the following reaction?

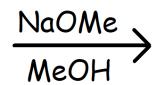


- 49. What is the name of the major elimination product of the reaction when 2-Bromo-3,4-dimethylhexane reacts with Methanol at 95° C?
- 52. What is the major product of this reaction?









54. What is the major product of this reaction?



56. Which of the following compounds will have the fastest reactivity in a SN2 reaction?

- A. Phenyl Bromide
- B. 1-Bromo-1-butene
- C. 1-Chloro-2-butene
- D. 3-lodo-1-Butene

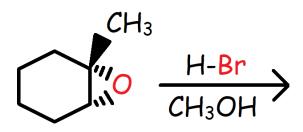
- 57. Which of the following bases will convert 1-Bromopentane to 1-Pentene with the greatest yield?
- A. NH₃
- B. N(CH₃)₃
- C. O-C CH₃
 CH₃
 CH₃
- D.

- 59. Which of the following statements is not true?
- A. Good leaving groups are weak bases and bad leaving groups are strong bases.
- B. The SN2 reaction works well with methyl and primary alkyl halides due to steric factors.
- C. The SN1 and E1 mechanism works very well with tertiary alkyl halides due to carbocation stability.
- D. Vinyl and aryl halides work very well with SN1 and SN2 reactions due to alkene stability.
- E. The E2 reaction works well with tertiary alkyl halides because the transition state resembles a more stable alkene.

- 58. What is the major product?
 - Br NaOCH₃ CH₃OH
- 60. Which of the following molecules is chiral?

Br

61. What is the major product of the reaction shown below?



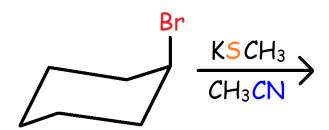
64. Draw the structure of the alkene that will produce Cyclopentanone and Acetone upon reacting with Ozone at -78° C followed by Zinc in Acetic Acid.

- 62. Which of the following statements is not true?
- A. Diastereomers have different melting points.
- B. Enantiomers have the same density but different optical properties.
- C. Meso compounds are optically inactive.
- D. Enantiomers can be separated by distillation.
- E. A racemic mixture of enantiomers will not rotate plane polarized light.

- 65. Which of the following is not an intermediate in the Ozonolysis reaction of Cis-3-hexene followed by DMS?
- A. Propanal
- B. Molozonide
- C. Carbonyl Oxide
- D. Ozonide
- E. Propanone

- 63. A solution has an enantiomeric excess of 60%. (R > S) How many grams of the R and S stereoisomers are present in the solution if a 50g sample of the chiral compound was dissolved in the original solution?
- 66. Which of the following reactions will generate a meso compound?
- A. Cis-2-butene and Br₂/CH₂Cl₂
- B. Cis-2-pentene + OsO₄ followed by NaHSO₃/H₂O
- C. Trans-2-butene with D₂ and Wilkinson's catalyst
- D. Trans-3-hexene with RCO₃H followed by H₃O⁺

67. What is the major product of the reaction shown below?



70. Which of the following reactions will generate the highest yield of Acetic Acid?

- A. 2-Butyne with O₃ followed by H₂O
- B. Propyne with KMnO₄ and H₃O⁺
- C. 2-Pentene with O₃ followed by Zn/CH₃COOH
- D. 2-Methyl-2-butene with O₃ followed by H₂O₂

68. Which of the following reagents will reacts with 2-Methylpropene to produce Carbon Dioxide?

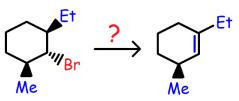
- (A) Ozone followed by (CH₃)₂S
- (B) KMnO₄, OH⁻, Cold, and Dilute
- (C) 1. OsO₄ 2. NaHSO₃/H₂O 3. HIO₄
- (D) KMnO₄, H₃O⁺, Warm, and Concentrated.

71. What is the major product of the reaction shown below?

69. Which of the following reagents will convert 1-Butyne to 2-Bromobutane?

- A. 1 eq. HBr in CH₂Cl₂
- B. 2 eq. Br₂ in CH₂Cl₂
- C. H₂ + Pd/BaSO₄/Quinoline followed by HBr/ROOR
- D. 1 eq. HBr/ROOR
- E. Na(s)/NH₃(I) followed by HBr in CH₂Cl₂

72. Which of the following compounds will be most effective in completing the reaction shown below?



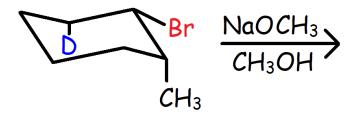
- A. NaOCH₃
- B. CH₃OH
- C. $KOC(CH_3)_3$
- D. KI
- E. NaSH

- 73. Which of the following reactions will produce Methyl Tert-butyl Ether with the greatest yield?
- 75. Which of the following reactions will not produce an alkene?

- A. $(CH_3)_3C-I + CH_3OH @ 20^0 C$
- B. (CH₃)₃C-Br + NaOCH₃ in CH₃OH
- C. CH_3 -I + $KOC(CH_3)_3$ in $(CH_3)_3COH$
- D. CH_3 -F + $KOC(CH_3)_3$ in $(CH_3)_3COH$

- A. Trans-1,2-dibromocyclohexane + KI in Acetone
- B. Tert-butanol with H₂SO₄ and heat
- C. 1-Methylcyclohexanol with H₃PO₄ and heat
- D. Cyclopentyl Bromide with NaN₃ in THF

74. What is the major product of this reaction?



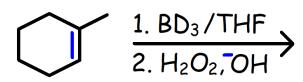
- 76. Which of the following alcohols will react most rapidly with H_2SO_4 ?
- A. 1-Pentanol

 C. 3- Methyl-4-hexen-3-ol

 B. 2-Pentanol

 D. 2-Phenyl-2-propanol
- 77. Which of the following reagents will convert 1-Methylcyclohexene to 6-Oxoheptanoic Acid?
- A. Ozone followed by DMS
- B. KMnO₄, Cold, OH⁻, and Dilute
- C. KMnO₄, Conc., OH⁻, and heat. followed by H₃O⁺
- D. MCPBA followed by H₃O⁺

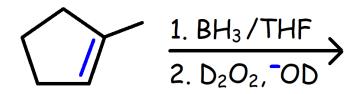
78. What is the major product of the reaction?



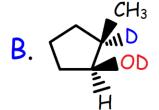
- A. CH3
- B. WOH
- C. CH₃
- D. CH

- 80. Which of the following alkyl halides will produce 2-Ethyl-1-pentene upon reacting with $KOC(CH_3)_3$ in $(CH_3)_3COH$?
- A. 1-Bromo-4-methylhexane
- B. 2-Bromo-4-methylhexane
- C. 3-Bromo-3-methylhexane
- D. 2-Bromo-3-methylhexane

79. What is the major product of the reaction?

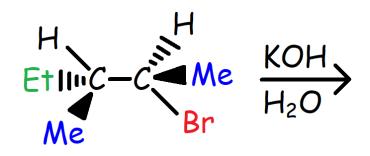


A. CH₃



- C. CH₃
- D. CH₃

81. What is the major product of the reaction shown below?



82. What is the major product?

84. What is the IUPAC nomenclature of the molecule shown below?

A. (2S, 3R)-2-Bromo-3-chloropentane

B. (2S, 3S)-2-Bromo-3-chloropentane

C. (2R, 3R)-2-Bromo-3-chloropentane

D. (2R, 3S)-2-Bromo-3-chloropentane

83. What is the major product that will form when (R)-2-Bromopentane is treated with KI in Acetone followed by NaCN in Acetonitrile?

85. What products will form from the reaction of 2-Pentyne with H_2 /Lindlar's catalyst followed by OsO_4 and $NaHSO_3/H_2O$?

Answers:

1. A

2. A

3. E

4. A

5. D

6. B

7. B

8. E

9. B

10. -30⁰

11. 8

12. 1-Bromo-1,2-dimethylcyclohexane

13. 1-Methoxy-1-methylcyclopentane

14. D

15. C

16. C

17. 60%

18. E

19. D

20. B

21. D

22. E

23. D

24. A

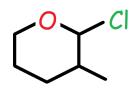
25. A

26. C

27. C

28. B

29.

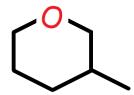


30. D

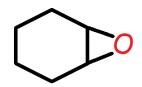
31.

- 32. C
- 33. 3 Products (3-Bromopentane, (R)-2-Bromopentane, and (S)-2-Bromopentane)
- 34. D
- 35. 1-Bromo-2-chloro-2-methylpropane
- 36. 3-Bromo-1-chloro-1-methylcyclohexane
- 37. E
- 38. D
- 39. A
- 40. C

41.



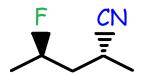
42.

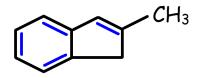


43.



- 44. C
- 45. 4
- 46. 2-Methyl-2-Pentene
- 47. 2-Methyl-1-Pentene
- 48. 1-Pentene
- 49. (E)-3,4-dimethyl-3-hexene
- 50. E
- 51.





53. C

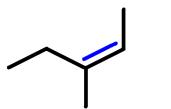
54. A

55. D

56. C

57. D (DBN)

58. (Z)-3-methyl-2-pentene



59. D

60. D

61.

62. D

63. 40g of the R-Isomer, 10g of the S-isomer

64.

65. E

66. D

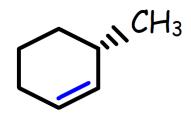
67.

68. D

69. E

70. A

71.



72. B

73. C

74. C

75. D

76. D

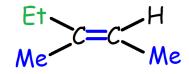
77. C

78. A

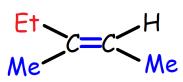
79. C

80. C

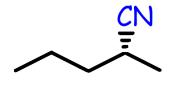
81.



82.



83.



84. A

85. A